

FUNDAMENTAL CONSTANTS (1986)

Avogadro's number, N	$6.02214 \times 10^{23} \text{ mol}^{-1}$	proton mass	1.00727647 amu
speed of light, c	$2.9979 \times 10^8 \text{ m/s}$	neutron mass	1.0086649 amu
Planck's constant, h	$6.62608 \times 10^{-34} \text{ J} \cdot \text{s}$	electron mass	$5.485799 \times 10^{-4} \text{ amu}$
gas constant, R	8.31451 J/mol·K	electron charge	$1.602177 \times 10^{-19} \text{ C}$
gas constant, R	0.082058 L·atm/mol·K	Faraday's constant, \mathcal{F}	$9.64853 \times 10^4 \text{ C/mol}$
gas constant, R	1.9872 cal/mol·K	Boltzmann's constant, k	$1.38066 \times 10^{-23} \text{ J/K}$

$$\Delta G^\circ = -RT \ln K$$

$$K_p = K_c (RT)^{\Delta n}$$

Selected Conversion Factors

$$1 \text{ atm} = 1.01325 \times 10^5 \text{ Pa} = 760 \text{ torr}$$

$$1 \text{ cal} \equiv 4.184 \text{ J}$$

$$1 \text{ in} \equiv 2.54 \text{ cm}$$

$$1 \text{ ft} \equiv 12 \text{ in}$$

$$1 \text{ mi} \equiv 5280 \text{ ft}$$

$$1 \text{ lb} = 453.592 \text{ g}$$